

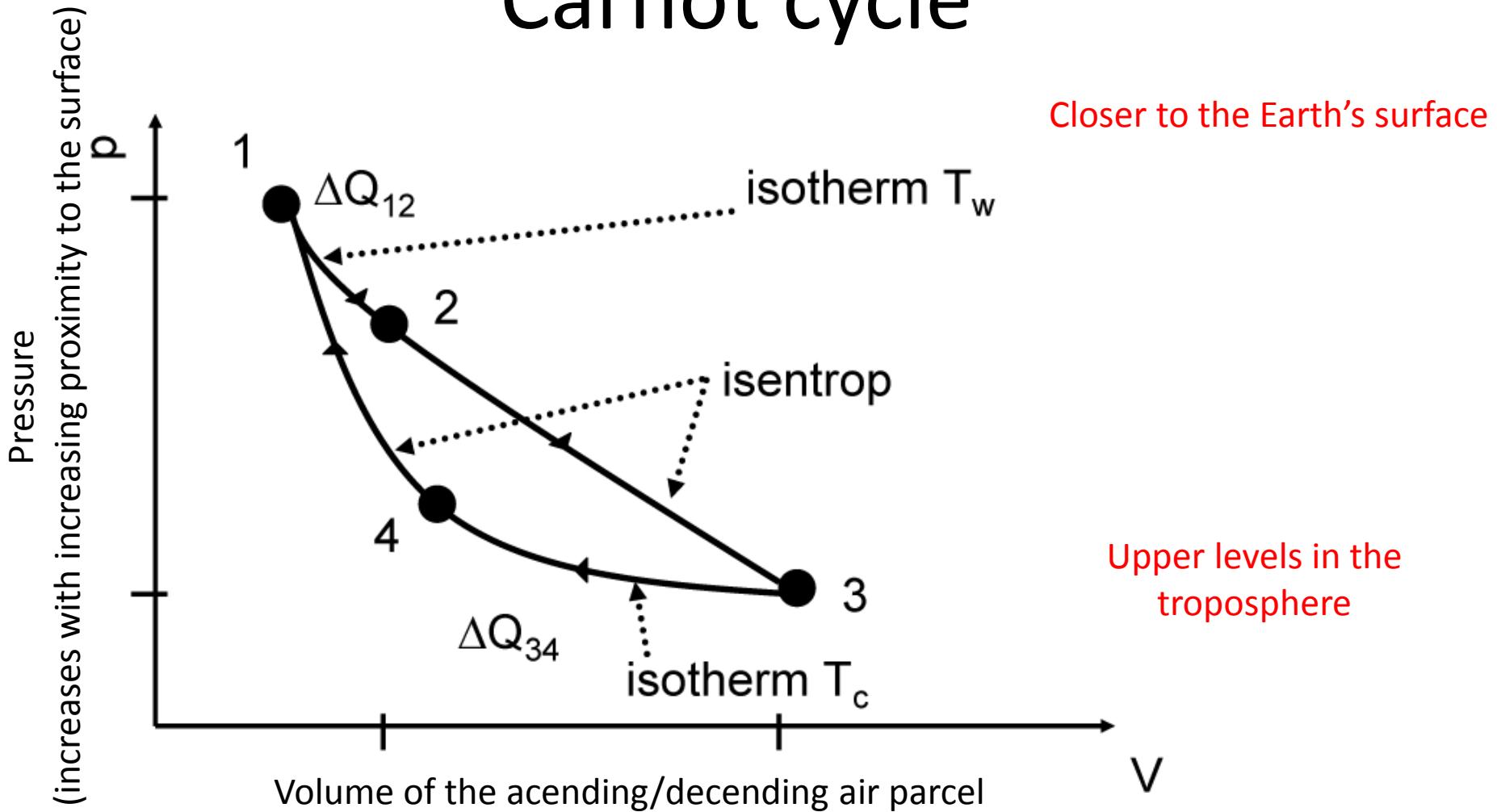
# Unit 4



Second Law of thermodynamics

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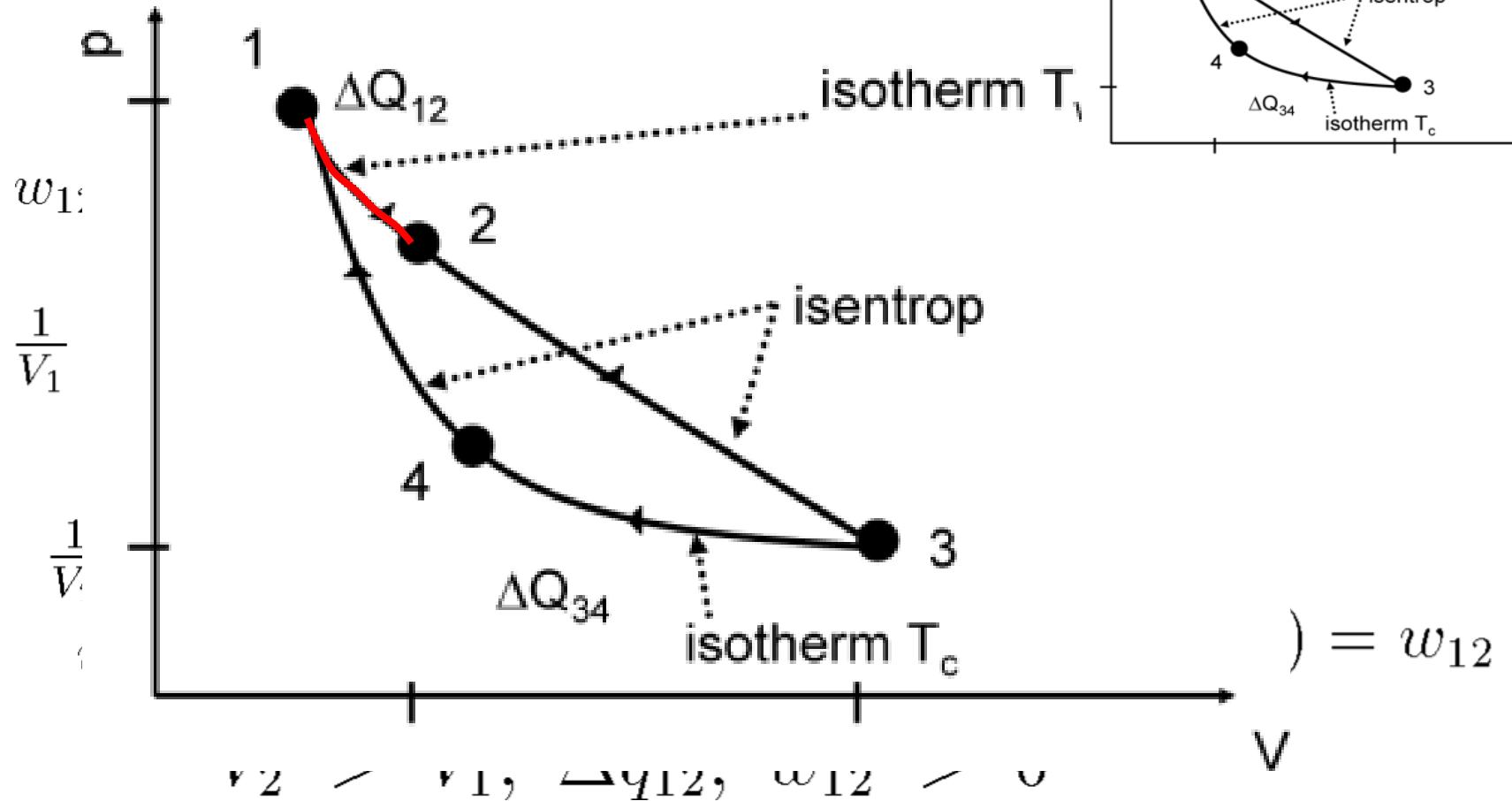
# Carnot cycle



$$\Theta_1 = T_w \left( \frac{1000}{p_1} \right)^{\frac{R_d}{c_p}} = T_c \left( \frac{1000}{p_4} \right)^{\frac{R_d}{c_p}}$$

$$\Theta_2 = T_c \left( \frac{1000}{p_3} \right)^{\frac{R_d}{c_p}} = T_w \left( \frac{1000}{p_2} \right)^{\frac{R_d}{c_p}}$$

# $1 \rightarrow 2$ : Reversible isothermal expansion



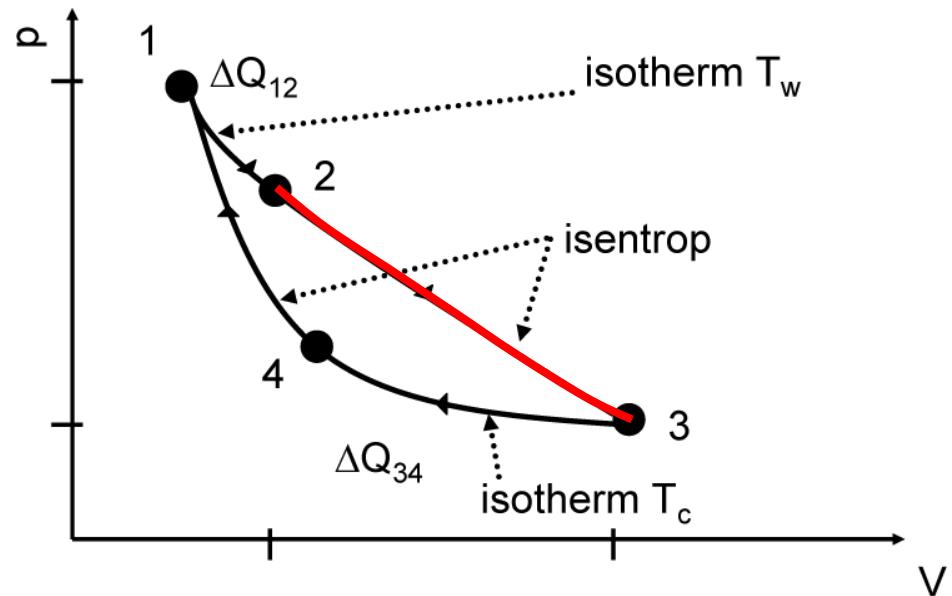
THM: heat can be added without change in T!

## $2 \rightarrow 3$ : Reversible adiabatic expansion

$$\Delta q_{23} = 0$$

$$\Delta u_{23} = -c_v(T_w - T_c)$$

$$w_{23} = -\Delta u_{23} = c_v(T_w - T_c)$$



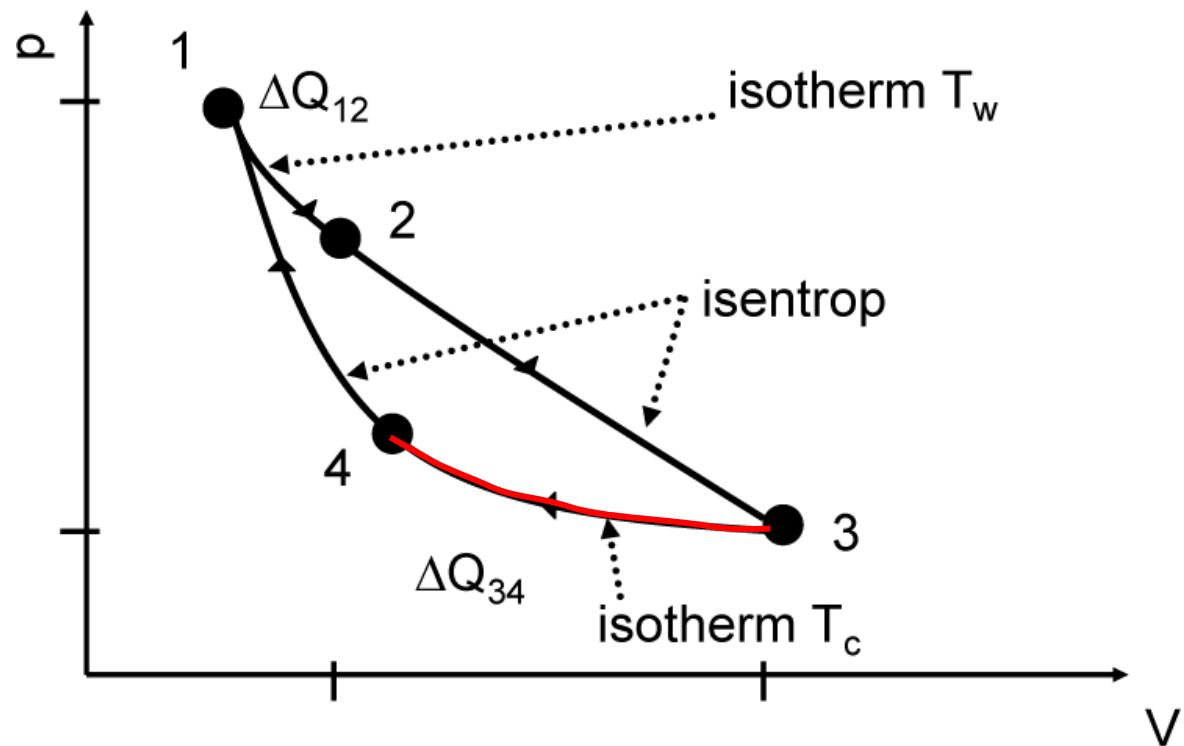
# $3 \rightarrow 4$ : Reversible isothermal compression

$$\Delta u_{34} = 0$$

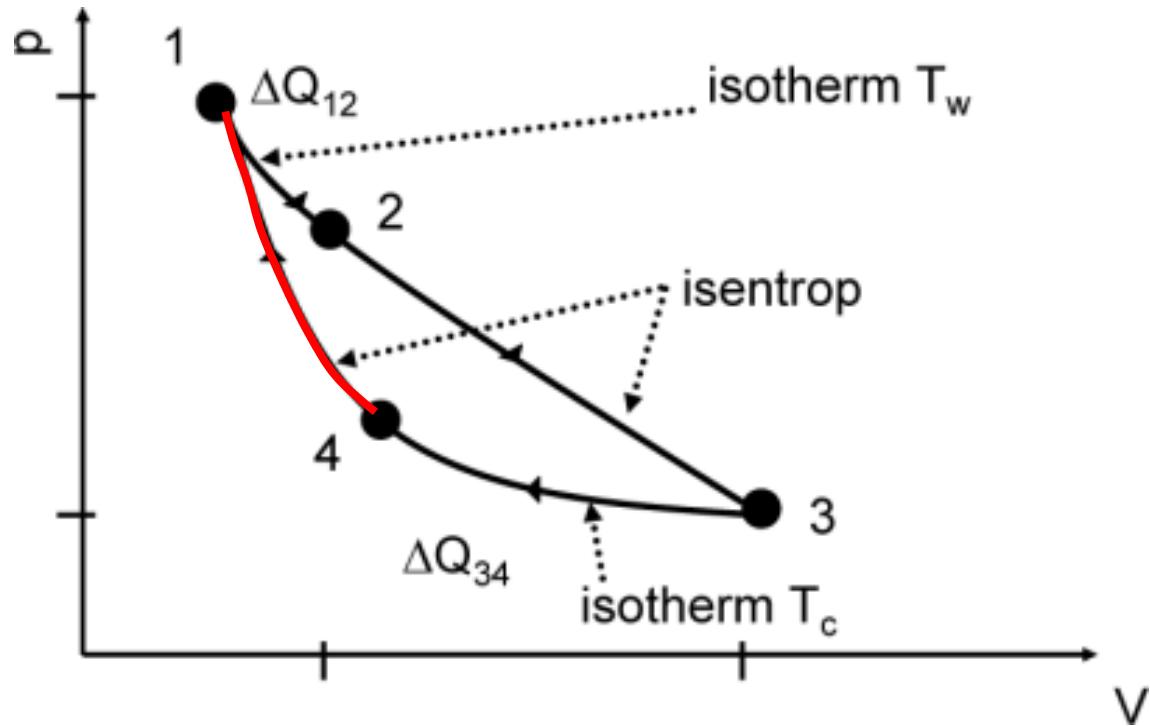
$$w_{34} = -R_d T_c \ln \frac{V_3}{V_4} = -R_d T_c \ln \left( \frac{p_1}{p_3} \left( \frac{T_c}{T_w} \right)^{\frac{c_p}{R_d}} \right)$$

$$q_{34} = -c_p T_c \left( \frac{\Theta_2}{\Theta_1} \right)$$

$$q_{34} = W_{34}$$



# $4 \rightarrow 1$ : Reversible adiabatic compression



$$q_{41} = 0$$

$$\Delta u_{41} = c_v(T_c - T_w) > 0$$

$$w_{41} = -\Delta u_{41} = -c_v(T_w - T_c) < 0$$

# Carnot cycle - THM

$$w = w_{12} + w_{23} + w_{34} + w_{41}$$

$$q = q_{12} + q_{23} + q_{34} + q_{41}$$

$$q = c_p(T_w - T_c) \ln\left(\frac{\Theta_2}{\Theta_1}\right)$$

$w = q_{12} + q_{34}$  and  $q_{34} < 0$  (

$$w = R_d(T_w - T_c) \ln(p_1 / p_3 (T_c / T_w)^{(cp/Rd)})$$

$$\eta = \frac{\Delta q_{12} + \Delta q_{34}}{\Delta q_{12}} = 1 + \frac{\Delta q_{34}}{\Delta q_{12}} = 1 - \frac{T_c}{T_w}$$

THM: Carnot-cycle has maximum possible efficiency  
of all processes between  $T_w$  and  $T_c$

# Thermodynamics potentials

Potential	Variables
$U(S, V, N)$	$S, V, N$
$H(S, P, N)$	$S, P, N$
$F(T, V, N)$	$V, T, N$
$G(T, P, N)$	$P, T, N$

$$dU = T dS - P dV + \mu dN$$

$$dH = T dS + V dP + \mu dN$$

$$dF = -S dT - P dV + \mu dN$$

$$dG = -S dT + V dP + \mu dN$$

Maxwell relation:  $\left(\frac{\partial T}{\partial V}\right)_{S,N} = -\left(\frac{\partial P}{\partial S}\right)_{V,N}, \dots$

# Example: isolated air parcel

$$dU(S, V, N) = TdS - PdV + \mu dN$$

$$dU(S, V, N) = \left( \frac{\partial U}{\partial S} \right)_{V,N} dS + \left( \frac{\partial U}{\partial V} \right)_{S,N} dV + \left( \frac{\partial U}{\partial N} \right)_{S,V} dN$$

$$\left( \frac{\partial U}{\partial S} \right)_{V,N} = T \quad \left( \frac{\partial U}{\partial V} \right)_{S,N} = -P \quad \left( \frac{\partial U}{\partial N} \right)_{S,V} = \mu$$